

# FEWG beyond SF6: Managing Halocarbon Emissions

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# Review of Progress to Date

- Industrial fugitive gases comprised 14% of the DOE Scope 1 and 2 GHG baseline 2008 inventory.
- 90% of baseline industrial fugitive GHG emissions were SF<sub>6</sub>.
- Between 2008 and 2010, DOE sites reduced industrial fugitive emissions by 39%.
- This effort has prevented the release of more than 600,000 metric tons of CO<sub>2</sub> equivalent since 2009!



(Fugitive Emissions Working Group Secretarial Achievement Award, photo courtesy of ANL)

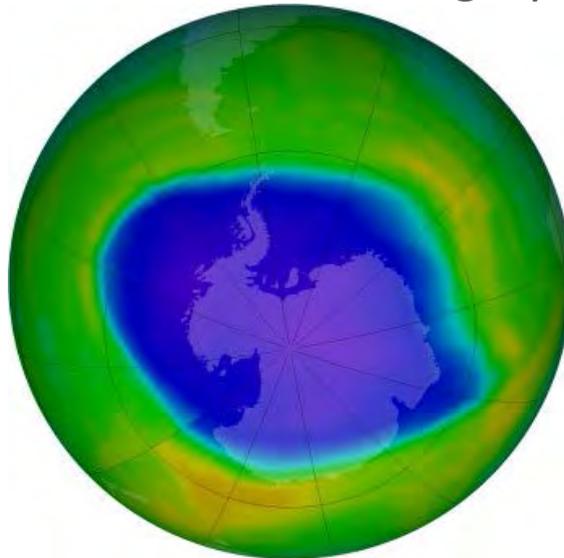
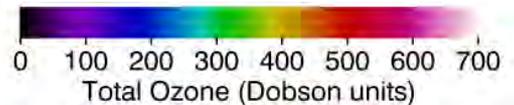
# What are halocarbons?

- Halocarbon compounds are chemicals in which one or more carbon atoms are linked with one or more halogen atoms (fluorine, chlorine, bromine or iodine).
- Halocarbons are typically nonreactive and nonflammable although some may be ionized.
  - Ionization via UV radiation can lead to ozone depletion.
  - Ionization via high temperature or electrical discharge is used in experimental high energy and nuclear physics.



# Halocarbon Emission Regulation

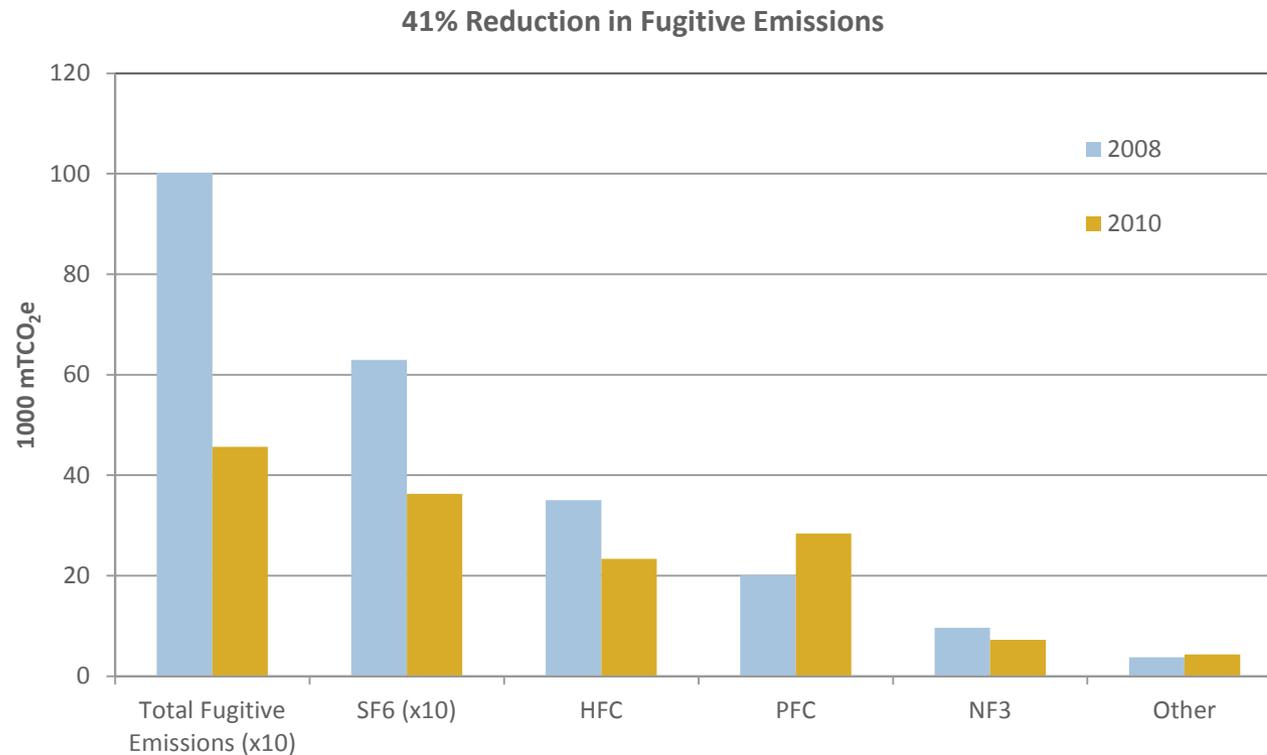
- Halocarbons (halogenated hydrocarbons) are covered under the Montreal Protocol.
  - The treaty is structured around several groups of halocarbons that have been shown to play a role in ozone depletion.
  - All of these ozone depleting substances (ODSs) contain either chlorine or bromine.
  - Fluorinate halocarbons are not covered (substances containing only fluorine do not contribute to ozone depletion).
- Aside from ODSs, fugitive emissions in this category are not otherwise regulated.



(Antarctic ozone hole, 2 Nov 2011, courtesy of NASA)

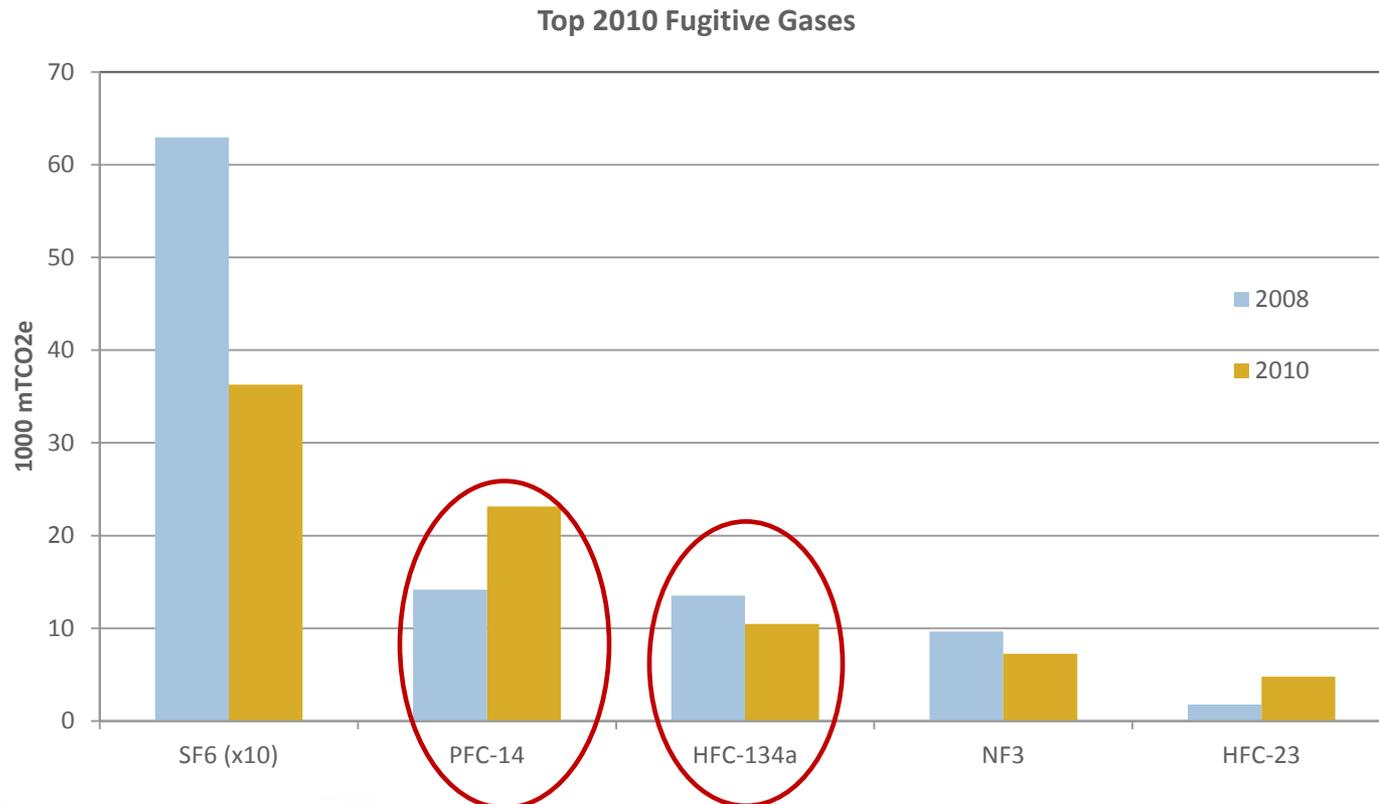
# DOE Emission Inventory for Fugitive Gases

- In 2008, fluorinated GHGs (HFCs and PFCs) comprised 7.2% of the fugitive inventory.
- In 2010:
  - HFCs decreased 33%
  - PFCs increased 42%



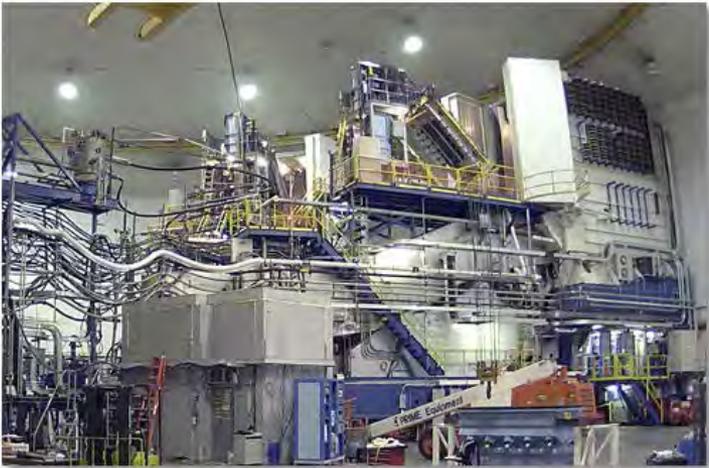
# DOE Emission Inventory for Halocarbons

- While SF<sub>6</sub> is clearly the fugitive gas with the largest impact to the DOE GHG inventory, 2 halocarbons used in high energy physics are among the top 5 fugitive gases.
  - CF<sub>4</sub> (PFC-14), and
  - R-134a (HFC-134a).



# Halocarbons on DOE Sites

- $\text{CF}_4$  (PFC-14) and R-134a (HFC-134a) are often used in ionization chamber detectors.
- Cherenkov (Čerenkov) detectors rely on “heavier” gases such as  $\text{C}_4\text{F}_{10}$  (PFC-3-1-10) or  $\text{C}_4\text{F}_8\text{O}$ .
  - $\text{C}_4\text{F}_{10}$  is currently tracked by the annual GHG inventory.
  - $\text{C}_4\text{F}_8\text{O}$  is currently not tracked by the GHG inventory although estimated GWP is 8,700.
- Hydrocarbons that are often mixed with halocarbons in ionization chambers include methane, ethane, and isobutane.
  - Aside from methane, these have very low GWPs.



(Hall A at JLab with Cherenkov detectors courtesy of JLab)



(Ring Imaging Cherenkov Detector at BNL courtesy of BNL)

# Challenges of Halocarbon Management

- Halocarbons have unique spectral and electronegative properties that are ideal for detectors.
- Generally these gases are used in equipment as mixtures.
  - As a result, there are significant challenges to safe and reliable processing of the spent gas.
  - Typical management practice is to vent gas after use.
- It is unclear from current inventory methods what percent is used as refrigerants, what percent is used in experiments, and what percent is for other uses.
  - Management of halocarbon refrigerants may focus more on leak detection and repair.
  - Management of halocarbons for experimental use may focus on reuse and recirculation.
- Successful recirculation projects exist.
  - FermiLab recirculates  $\text{CF}_4$  at a collider experiment.
  - CERN successfully recirculates R-134a.
  - Any other success stories?



# For FEWG Discussion

- Some sites are already actively managing halocarbons to minimize emissions.
  - Are there opportunities to address this issue at other sites?
  - Could managing halocarbons limit ongoing efforts to manage SF<sub>6</sub>?
- Most halocarbons are covered in the GHG inventory.
  - Should C<sub>4</sub>F<sub>10</sub> be added to the inventory?
- Are hydrocarbons (ethane, isobutane) associated with ionization detection significant enough to be added to the inventory?
- Are there other gases that are not currently tracked that should be tracked?



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